

Domino C-F Bond Activation of the CF₃ Group: Synthesis of Fluorinated Dibenzo[a,c][7]annulenes from 2-(Trifluoromethyl)-1alkenes and 2,2'-Diceriobiaryls

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Supporting Information

ABSTRACT: The construction of ring-fluorinated sevenmembered carbocycles was readily achieved via the domino S_N2'-type/S_NV reaction between 2-(trifluoromethyl)-1-alkenes and 1,4-carbodianions. The S_N2'-type reaction of 2-(trifluoromethyl)-1-alkenes with 2,2'-diceriobiaryls generated the intermediary 1,1-difluoro-1-alkenes bearing a monoceriobiaryl moiety, which in turn underwent intramolecular S_NV reaction to afford fluorinated 5*H*-dibenzo [a,c] [7] annulenes.

 $^{\mathsf{T}}$ arbon-fluorine (C-F) bonds are generally considered to → be difficult to cleave because of their high bond energy, whereas 2-(trifluoromethyl)-1-alkenes and 1,1-difluoro-1-alkenes exhibit unique reactivities in defluorinative additionelimination processes. Nucleophilic attack to 2-(trifluoromethyl)-1-alkenes typically proceeds at the carbon atom γ to the fluorine substituents and is followed by fluoride elimination to afford 1,1-difluoro-1-alkenes (S_N2'-type reaction, Scheme 1, eq

Scheme 1. Allylic and Vinylic C-F Bond Activation in S_N2'-Type and S_NV Reactions

S_N2'-type reaction

$$F_{3}C \xrightarrow{R} \xrightarrow{\overline{N}U} F_{2}C \xrightarrow{R} F_{2}C \xrightarrow{N} N_{U}$$

Domino S_N2'-type / S_NV reaction

$$F_{3}C \xrightarrow{+} \underbrace{S_{N}2'\text{-}type}_{-F^{-}} \quad \begin{bmatrix} R \\ F_{2}C \\ Nu^{2} & Nu^{1} \end{bmatrix} \xrightarrow{-F^{-}} F_{Nu^{2}} \xrightarrow{Nu^{1}} (3)$$

1).^{1,2} On the other hand, 1,1-difluoro-1-alkenes are susceptible to nucleophilic substitution at the carbon atom α to the fluorine substituents to give monofluoroalkenes (S_NV reaction, Scheme 1, eq 2).^{1,3}

Thus, appropriate choice of binucleophiles enables straightforward ring construction via double activation of C-F bonds by the sequential $S_N 2'$ -type/ $S_N V$ process (Scheme 1, eq 3). This protocol enables selective activation of two C-F bonds in a CF₃ group and retention of one C-F bond, which has long been considered troublesome because of the shielding effect of the CF₃ group.⁴ In fact, we have reported 2-fluoroquinoline synthesis from 2-(trifluoromethyl)-1-alkenes via the S_N2'-type/ S_NV sequence, initiated by treatment with *ortho*-lithiated aniline derivatives.⁵ Similarly, we have also achieved 3-fluoropyrazole synthesis via the formal [3 + 2]-cyclization between 2-(trifluoromethyl)-1-alkenes and hydrazines, in which the ring closure was not completed by a simple S_NV reaction.⁶ Most recently, following our results, Xiao and Zhang et al. reported a one-pot synthesis of 2-fluoro-4H-pyrans in which 1,3dicarbonyl compounds were used as binucleophiles in the presence of K₂CO₂.

To take full advantage of the sequence, we embarked on the construction of seven-membered carbocycles by using binucleophiles containing two carbanion moieties (Scheme 2). As a result, we succeeded in a one-pot synthesis of 7-fluoro-5H-dibenzo[a,c][7]annulenes⁸ by treatment of 2-(trifluoromethyl)-1-alkenes with 2,2'-dimetallobiaryls. The cycloheptatriene core is widely used in bioactive agents 10 and functional materials. 11 In spite of their utility, their conventional synthetic methods typically required harsh conditions, expensive

Scheme 2. Seven-Membered Carbocycle Construction via Domino S_N2'-Type/S_NV Reactions

$$F_{3}C$$

$$+$$

$$M$$

$$S_{N}2'-type$$

$$-F^{-}$$

$$M$$

$$S_{N}V$$

$$-F^{-}$$

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Table 1. Screening of Conditions for Domino S_N2'-Type/S_NV Reaction between 1a and 2a

entry	MXn	solvent	conditions	3aa ^a (%)	4aa ^a (%)
1		THF	55 °C, 4 h	58	27
2	$MgBr_2$	THF	55 °C, 4 h	ND^{b}	ND^{b}
3	$ZnCl_2$	THF	55 °C, 4 h	N.D. ^b	ND^{b}
4	CuBr	THF	55 °C, 4 h	ND^{b}	ND^{b}
5	$CeCl_3$	THF	55 °C, 4 h	82	7
6	$CeCl_3$	Et ₂ O	40 °C, 5 h	71	16
7	$CeCl_3$	1,4-dioxane	55 °C, 6 h	29	22
8	$CeCl_3$	THF	-78 °C, 15 min then 55 °C, 4 h	40	trace
9	$CeCl_3$	THF	0 °C, 15 min then 55 °C, 4 h	37	12
10	$CeCl_3$	THF	rt, 15 min then 55 °C, 4 h	91 (90)	6

^aYield was determined by ¹⁹F NMR measurement using PhCF₃ as an internal standard. Isolated yield is given in parentheses. ^bND = not detected.

reagents, and/or multiple steps.⁸ In contrast, our approach enables a short-step synthesis of dibenzo[7]annulenes by using readily available substrates and reagents.

First, we sought suitable conditions for the domino S_N2'type/ S_NV reaction using α -(trifluoromethyl)styrene (1a) and 2,2'-dibromobiphenyl (2a) as model substrates (Table 1). Dibromobiphenyl 2a was treated with 2.1 equiv of n-BuLi and N,N,N',N'-tetramethylethylenediamine (TMEDA) in THF at room temperature for 15 min. Then, the mixture was reacted with (trifluoromethyl)styrene 1a at 55 °C for 4 h to give dibenzo[7] annulene 3aa in 58% yield, as well as difluoroalkene 4aa, formed only by the S_N2'-type reaction, in 27% yield (entry 1). To improve the reactivities of the intermediary di- and monometallobiphenyls, several other metallic species were examined (entries 2-5). When MgBr₂, ZnCl₂, or CuBr was employed for transmetalation, the reaction was seriously inhibited (entries 2-4). However, the use of CeCl₃, which would generate 2,2'-diceriobiphenyl, drastically promoted the domino S_N2'-type/S_NV reaction to afford 3aa in 82% yield, suppressing the formation of 4aa (entry 5).¹² Since cerium reagents generally exhibit reduced basicity along with substantial nucleophilicity, the reaction of 2,2'-diceriobiphenyl with 1a seems to proceed as expected, suppressing deprotonation of the solvent. Although the solvent effects of other ether solvents, such as diethyl ether and 1,4-dioxane, were examined, THF served as the best medium (entries 6 and 7). For further improvement in the yield of 3aa, we changed the temperature at the initial stage of the reaction between 1a and the diceriobiphenyl, as that is less stable to heating (entries 8-10). After addition of 1a, stirring at −78 or 0 °C for 15 min followed by raising the temperature to 55 °C retarded the reaction (entries 8 and 9). Finally, when the temperature was kept at room temperature for 15 min to consume the diceriobiphenyl and then raised to 55 °C, 3aa was obtained in 90% isolated yield (entry 10).

With the optimized conditions in hand, we have investigated the scope of 2-(trifluoromethyl)-1-alkenes 1 and 2,2'-dibromobiaryls 2 (Table 2). Trifluoromethylstyrenes 1b and 1c, bearing a methyl group on the phenyl group, effectively underwent the domino reaction with the in situ generated diceriobiphenyl derived from 2a, regardless of the position of

Table 2. Synthesis of Dibenzo[7]annulenes 3

entry	1 , R	x (equiv)	2	3 (yield, 4 %)
1	1a, Ph	1.2	2a	3aa (90)
2	1b , 4-MeC ₆ H ₅	1.3	2a	3ba (77)
3	1c, 3-MeC ₆ H ₅	1.2	2a	3ca (72)
4	1d, 4-FC ₆ H ₅	1.2	2a	3da (78)
5	1e, 3-FC ₆ H ₅	1.4	2a	3ea (80)
6	1f, 4-ClC ₆ H ₅	1.1	2a	3fa (83)
7	1g, 3-ClC ₆ H ₅	1.4	2a	3ga (80)
8 ^b	1h , 4-MeOC ₆ H ₅	1.0	2a	3ha (44)
9 ^b	1i, 3-MeOC ₆ H ₅	1.2	2a	3ia (44)
10 ^b	1j, 4-CF ₃ C ₆ H ₅	1.2	2a	3ja (57)
11 ^b	1k, SiMe ₂ Ph	1.4	2a	3ka (79)
12	1a , Ph	1.2	2b	3ab (80)
13 ^b	1k, SiMe ₂ Ph	1.4	2b	3kb (72)
14	1a, Ph	1.2	2c	3ac (80)
15 ^b	1k, SiMe ₂ Ph	2.0	2c	3kc (70)

 a Isolated yield. b BF $_3$ ·OEt $_2$ (2.0 equiv) was added 5 min after addition of 1. After being stirred at room temperature for 10 min, the mixture was heated to 55 $^\circ$ C.

the substituent, leading to the corresponding dibenzo[7]-annulenes 3ba and 3ca in 77% and 72% yields, respectively (entries 2 and 3). Reactions of (trifluoromethyl)styrenes 1d-g bearing a fluorine or chlorine substituent also proceeded without loss of the C–F or C–Cl bond on the benzene ring (entries 4-7). Methoxy-bearing (trifluoromethyl)styrenes 1b and 1i, (trifluoromethyl)styrene 1b bearing another CF $_3$ group, and 2-silylated trifluoropropene 1b participated in the reaction with the aid of $BF_3 \cdot Et_2O$ (entries 8-11). Not only 2a but also di-tert-butylated and tetramethylated biphenyls 2b and 2c underwent diceriation followed by the domino $5n^2$ -type/ $5n^2$

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reaction with (trifluoromethyl)alkenes 1a and 1k to afford the corresponding dibenzo [7] annulenes 3ab, 3kb, 3ac, and 3kc in high yields (entries 12–15).

Unambiguous structural characterization of dibenzo[7]-annulenes 3 was achieved by X-ray crystallographic analysis of 3fa (Figure 1). The bond lengths of C5–C6 and C6–C7 are

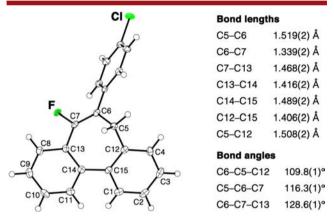


Figure 1. ORTEP drawing of dibenzo[7]annulene **3fa** with 50% ellipsoid probability.

1.519(2) and 1.339(2) Å, respectively. Moreover, the bond angles of C6–C5–C12, C5–C6–C7, and C6–C7–C13 are 109.8(1)°, 116.3(1)°, and 128.6(1)°, respectively. These data indicate that the hybridization modes of C5, C6, and C7 are sp³, sp², and sp², respectively, and C6–C7 is a double bond. Thus, in the domino $S_{\rm N}2'$ -type/ $S_{\rm N}V$ reaction of trifluoromethylalkenes 1 with diceriobiaryls, the latter $S_{\rm N}V$ reaction proceeded without migration of C–C double bonds.

Further transformations of dibenzo[7]annulene 3ka bearing a silyl group were achieved by treatment with electrophiles (Scheme 3, eqs 1 and 2). Desilylation of 3ka proceeded by

Scheme 3. Chemical Transformation of 3ka

addition of tetrabutylammonium fluoride in THF to afford protodesilylation product $\mathbf{5}$ in 87% yield (Scheme 3, eq 1). In addition, $3\mathbf{ka}$ also underwent electrophilic substitution with Br_2 in CH_2Cl_2 to give brominated dibenzo[7]annulene $\mathbf{6}$ in 81% yield (Scheme 3, eq 2). Bromo[7]annulene $\mathbf{6}$ serves as a platform for further functionalized cycloheptatrienes via transition-metal-catalyzed coupling reactions.

In conclusion, we have established a one-pot synthesis of fluorinated dibenzo[7]annulenes that involves sequential activation of two C–F bonds of the CF₃ group in 2-(trifluoromethyl)-1-alkenes. Our current protocol provides ready access to valuable cycloheptatriene derivatives. Further-

more, the obtained fluorine-containing dibenzo[7]annulenes are less accessible by other methods. 13

ASSOCIATED CONTENT

S Supporting Information

The Supporting Information is available free of charge on the ACS Publications website at DOI: 10.1021/acs.orglett.6b03743.

Experimental details, characterization data, and NMR spectra (PDF)

X-ray crystallographic data of 3fa (CIF)

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Notes

The authors declare no competing financial interest.

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